

PRODUCTS OF THE CONDENSATION OF 1,2,3-
TRIMETHYLBENZIMIDAZOLIUM SALTS WITH
o-HYDROXY ALDEHYDES

L. S. Éfros, É. R. Zakhs,
and N. K. Beresneva

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The neutralized products of the condensation of 1,2-dimethylbenzoxazolium salts [1] and 1,2-dimethylbenzothiazolium salts [2] with o-hydroxyaldehydes exist in the form of colorless spiroyrans (A, Z = O, S) which are converted into labile colored forms (B, Z = O, S) merely on heating or by the action of light.

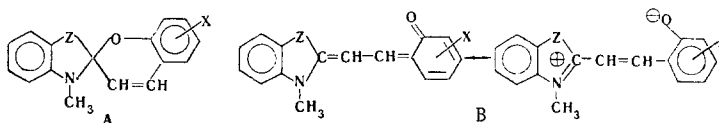
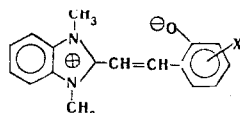
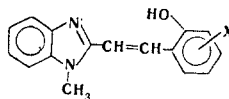


TABLE 1



Compound	X	mp, °C (decomp)	Empirical formula	N, %		UV Spectra		Yield, %
				found	calc.	λ_{max}, nm ($\epsilon \cdot 10^{-4}$)	$c \cdot 10^5 M$	
I	H	144—147 (water)	C ₁₇ H ₁₆ N ₂ O	10,6	10,6	320(1,6); 455(0,48)	6,4	62
II	5-NO ₂	291—296 (ethanol)	C ₁₇ H ₁₅ N ₃ O ₃	13,4	13,6	260(1,08); 300(1,31); 420(1,96)	4,8	83
III	3,5-Br ₂	293—295 (ethanol)	C ₁₇ H ₁₄ Br ₂ N ₂ O	6,6	6,6	248(1,89); 320(1,57); 450(1,28)	4,4	79
IV	5,6-Benzo	290 (benzene)	C ₂₁ H ₁₈ N ₂ O	8,8	8,9	248(3,04); 283(1,85); 347(1,08); 490(1,68)	6,5	77

TABLE 2



Compound	X	mp, °C (decomp.)	Empirical formula	N, %		Yield, %
				found	calc.	
V	H	239 (ethanol)	C ₁₆ H ₁₄ N ₂ O	11,2	11,2	82
VI	5-NO ₂	240 (50% ethanol)	C ₁₆ H ₁₃ N ₃ O ₃	14,4	14,2	98
VII	3,5-Br ₂	304 (ethanol)	C ₁₆ H ₁₂ Br ₂ N ₂ O	7,1	6,8	87

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In contrast to this, the stable form of the analogous 1,2,3-trimethylbenzimidazolium derivatives that we have obtained (I-IV) is the colored open structure (B, Z = N-CH₃), which cannot be converted into the colorless form A. We explain this difference by the low electrophilicity of position 2 of the imidazole ring. Since the IR spectra of I-IV (in KBr) lack absorption in the 1650-1690 cm⁻¹ region, it may be assumed that the structure of these compounds largely corresponds to the bipolar limiting structure (B).

1,3-Dimethyl-2-(2-oxidoarylvinyl)benzimidazolium (I) and Its Derivatives (II-IV). These were obtained by the brief heating of equimolar amounts of 1,2,3-trimethylbenzimidazolium methosulfate and the corresponding o-hydroxyaldehydes in ethanol in the presence of piperidine (Table 1).

To confirm their structure, compounds I-III were also obtained by the action of ammonia on the methosulfates of the corresponding 1,3-dimethyl-2-(2-hydroxy-X-styryl)benzimidazoles, obtained in their turn by the quaternization of 1-methyl-2-(2-hydroxy-X-styryl)benzimidazoles (V-VII), which were prepared by Horwitz's method [3] (Table 2). The C and H contents of IV and V and the Br contents of III and VII were determined; the experimental results agreed with the calculated figures.

LITERATURE CITED

1. Australian Patent No. 62,296 (1960); British Patent No. 889,186 (1962).
2. R. Guglielmetti, E. Pretelli, and E. J. Metzger, *Bull. Soc. Chim. France*, 1967, 2824.
3. L. Horwitz, *J. Am. Chem. Soc.*, 77, 1687 (1955).